

Chemical Bonding In Solids And Fluids (Ellis Horwood Series In Physical Chemistry) By Mark Ladd

By Mark Ladd

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Chemical Bonding in Solids examines how atoms in solids are bound together and how this determines the structure and properties of materials. <http://www.amazon.com/Chemical-Bonding-Solids-Inorganic-Chemistry/dp/0195089928>

Chemical bonding between atoms results in compounds that can be very different from the parent atoms. If we consider a solid crystal of sodium chloride, <http://www.visionlearning.com/en/library/Chemistry/1/Chemical-Bonding/55>

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or when covalent bonds extend in networks through solids that are not composed of discrete Early speculations into the nature of the chemical bond, http://en.wikipedia.org/wiki/Chemical_bond

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In ionic and molecular solids, there are no chemical bonds between the molecules, atoms, or ions. This type of chemical bonding is called metallic bonding. http://chemwiki.ucdavis.edu/?title=Textbook_Maps/General_Chemistry_Textbook_Maps/Map:_Brown,_LeMay,_%26_Bursten_%22Chemistry:_The_Central_Science%22/11.Liquids_and_Intermolecular_Forces/11.8:_Bonding_in_Solids

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(said he) got a strange opinion that every thing should be taught by
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